

PRODUCTION OF ACETIC ACID FROM VINE SHOOTS VIA PHOSPHORIC ACID-CATALYZED STEAM EXPLOSION AND FERMENTATION



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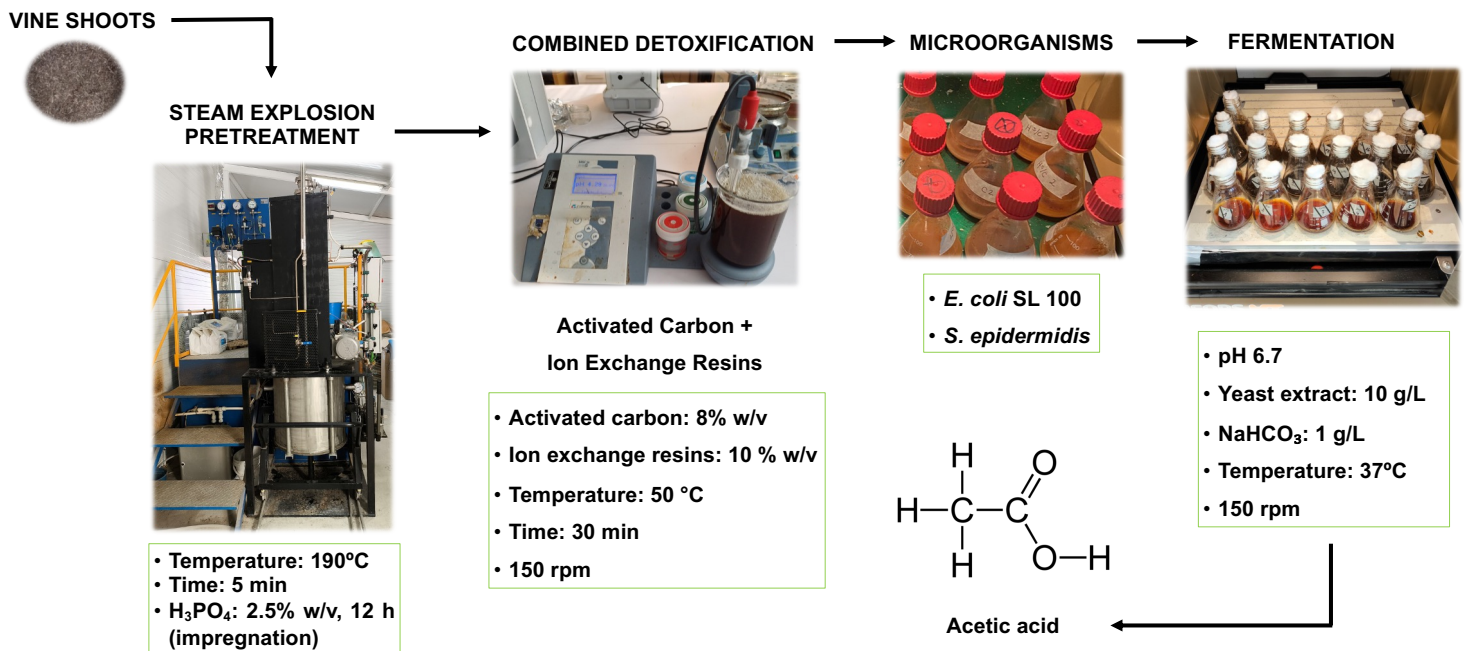
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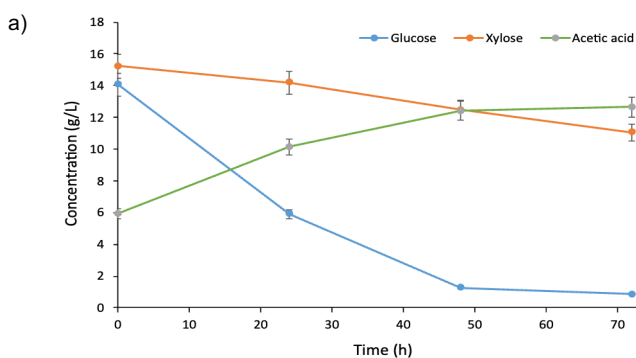
INTRODUCTION

To support a circular bioeconomy, this study explores the bioconversion of vine shoots, an abundant lignocellulosic residue, into acetic acid. The process involves phosphoric acid-catalyzed steam explosion pretreatment and detoxification, followed by fermentation with *E. coli* and *S. epidermidis* strains.

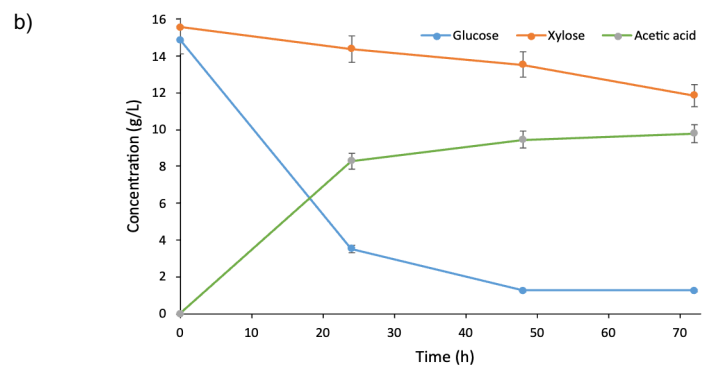
MATERIAL AND METHODS



RESULTS AND DISCUSSION



Regarding Figure a, *S. epidermidis* reached the highest total acetic acid concentration at 12.7 g/L. However, its efficiency was lower than it appeared, resulting in a yield of 0.43 g/g. This is because the strain benefited from approximately 5.9 g/L of acetate that was already present in the medium before fermentation began.



As shown in Figure b, *E. coli* SL100 displayed superior bioconversion performance in terms of net production. This strain synthesized 9.8 g/L of acetic acid from 17.3 g/L of metabolized sugars, achieving a higher yield of 0.56 g/g. These results highlight the potential of the SL100 strain, as its efficiency is comparable to that of recombinant strains used on other corn-based hydrolysates.

CONCLUSIONS

- Phosphoric acid-catalyzed steam explosion successfully breaks down the resistance of vine shoots, making them suitable for fermentation.

- While *S. epidermidis* is robust in accumulating the final product, *E. coli* SL100 is more efficient at converting sugar into acid.
- Due to its higher conversion yield, *E. coli* SL100 is identified as the best candidate for large-scale industrial production.
- This process provides a validated method to turn viticulture waste into a renewable and competitive alternative to fossil-based acetic acid.

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