

Circular carbon and plastic waste valorization via photoelectrochemical H₂ production and CO₂ conversion

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The continuous rise in atmospheric carbon dioxide (CO₂) concentration is one of the most critical drivers of global climate change. Recent measurements indicate that the global average atmospheric CO₂ concentration exceeded 424 ppm in 2024 and is projected to reach approximately 426–430 ppm during 2025, representing the highest level recorded in at least two million years [1]. Compared with the pre-industrial concentration of about 280 ppm, this represents an increase of more than 50%, largely driven by fossil fuel combustion, industrial activities, and land-use change. In parallel with rising CO₂ emissions, plastic waste has become one of the most severe environmental pollution challenges. Global plastic production has exceeded 400 million tons per year, and only a small fraction of this material is effectively recycled. Most plastic waste accumulates in landfills or the natural environment. According to Kim et al. [2], among the different polymer types, polyethylene (PE), polypropylene (PP), polyethylene terephthalate (PET), polyvinyl chloride (PVC), and polystyrene (PS) represent the dominant fraction of global plastic waste streams. Polyolefins such as PE and PP alone account for nearly 50% of global plastic waste, making them prime candidates for valorization technologies.

In recent years, carbon capture and utilization (CCU) strategies have been proposed to transform CO₂ from an environmental liability into a valuable feedstock for fuels and chemicals. Among the possible products, methanol and formic acid have attracted particular attention. Methanol is a versatile platform molecule used as fuel, H₂ carrier, and feedstock for the chemical industry, while formic acid can serve as a H₂ storage medium and chemical intermediate [3], [4]. Moreover, photoelectrochemical (PEC) technologies represent a promising approach for the conversion of CO₂ using renewable energy. By combining light absorption, charge separation, and catalytic reactions within integrated systems, PEC processes can convert solar energy directly into chemical fuels. Nevertheless, the efficiency of CO₂ reduction remains limited. Current electrochemical and photoelectrochemical systems typically achieve Faradaic efficiencies ranging from about 20–60% for formic acid, depending on the catalyst and electrolyte conditions, while the production of methanol generally remains lower due to the complexity of the multi-electron transfer reaction [5], [6].

Another major limitation of CO₂ conversion pathways is the availability of sustainable H₂. Conventional production via water electrolysis or steam methane reforming requires significant energy inputs. Recently, PEC plastic waste reforming has emerged as a promising strategy to simultaneously degrade plastic polymers and generate H₂. In these systems, plastic substrates act as electron donors, enabling H₂ evolution while producing valuable oxidation products. Then, integrating plastic waste upcycling with CO₂ conversion technologies offers a promising pathway to simultaneously address plastic pollution, renewable H₂ production, and carbon utilization. This research proposes an integrated approach that couples photoelectrochemical plastic waste upcycling for H₂ production with CO₂ hydrogenation to methanol and formic acid, thereby establishing a synergistic pathway for the valorization of both carbon emissions and solid plastic residues.

The specific objectives of the study include *i) design and development of photoelectrochemical catalytic systems* capable of reducing CO₂ into value-added products such as methanol and formic acid under solar-driven conditions, *ii) investigation of plastic waste upcycling pathways through photoelectrochemical oxidation processes* to produce H₂ while transforming polymers into valuable chemical intermediates, *iii) integration of H₂ generation and CO₂ reduction processes* in order to establish a coupled system for solar-driven fuel and chemical production, *iv) evaluation of energy efficiency and environmental performance*, including sustainability metrics such as energy balance, carbon utilization efficiency, and potential greenhouse gas emission reductions, and *v) assessment of scalability and feasibility*, considering process integration strategies that could contribute to decentralized and renewable fuel production systems.

The proposed research will focus on the development and evaluation of integrated photoelectrochemical systems combining CO₂ reduction and plastic waste upcycling. The methodology will include three main components: catalyst development, system integration, and sustainability assessment. Advanced catalytic materials will be investigated for the selective reduction of CO₂ to methanol and formic acid. Particular attention will be

given to nanostructured catalysts and semiconductor photoelectrodes designed to enhance light absorption, charge separation, and catalytic selectivity.

The performance of these systems will be evaluated through key electrochemical parameters including: i) Faradaic efficiency, ii) current density, iii) product selectivity, and iv) solar-to-fuel efficiency. These indicators are commonly used to assess the effectiveness of photoelectrochemical fuel production systems. As for the plastic waste upcycling for H₂ production, the anodic reaction of the photoelectrochemical system will involve the oxidation of plastic waste streams. These polymers represent a large fraction of global plastic waste and are therefore promising feedstocks for plastic upcycling technologies. As depicted by Li et al. [7], through photoelectrochemical oxidation processes, these polymers can be converted into smaller organic molecules while simultaneously generating protons and electrons that drive hydrogen evolution at the cathode. The H₂ fraction generated in this step can subsequently be used to facilitate the hydrogenation of CO₂ into methanol and formic acid.

Regarding the techno-economic and sustainability assessment, a comprehensive sustainability assessment will be conducted using process modeling and techno-economic analysis (TEA). The performance indicators on each dimension to be considered are listed in **Table 1**. These metrics will allow the identification of key technological bottlenecks and will provide insights into the scalability of integrated photoelectrochemical systems for renewable fuel production [8].

Table 1. Indicators evaluated in each dimension.

Technical indicators	Economic indicators	Environmental indicators
1. CO ₂ conversion efficiency.	1. Levelized cost of methanol (LCOM).	1. Life-cycle greenhouse gas emissions.
2. H ₂ production rate.	2. Levelized cost of hydrogen (LCOH).	2. Energy returns on energy invested (EROI).
3. Process mass intensity.	3. Capital expenditure (CAPEX).	
4. Solar-to-fuel efficiency.	4. Operational expenditure (OPEX).	
5. Carbon utilization efficiency.	5. Payback period.	

The proposed research is expected to contribute to the development of innovative technologies for the sustainable production of fuels and chemicals. By combining CO₂ utilization with plastic waste upcycling, the proposed approach aims to establish a synergistic pathway for addressing two major environmental challenges simultaneously. The integration of photoelectrochemical processes could enable the production of solar fuels such as methanol while reducing reliance on fossil resources. Methanol produced from captured CO₂ supports the transition toward a low-carbon energy system. Furthermore, the use of plastic waste as a feedstock for H₂ production represents a promising strategy for waste valorization. Instead of being disposed of or incinerated, plastic materials could be transformed into valuable chemical intermediates while contributing to the generation of renewable fuels. From a broader perspective, this research supports the development of circular carbon technologies, where carbon atoms are continuously recycled rather than emitted into the atmosphere. The integration of solar energy, CO₂ utilization, and waste upcycling aligns with global efforts to promote sustainable energy systems and resource efficiency.

This work proposes an integrated photoelectrochemical strategy for the simultaneous valorization of carbon dioxide and plastic waste. By coupling H₂ production from plastic waste upcycling with CO₂ reduction to methanol and formic acid, the proposed system seeks to develop a sustainable pathway for renewable fuel production. The combination of advanced catalytic materials, solar-driven electrochemical processes, and sustainability assessment will provide insights into the feasibility and potential impact of these technologies. Ultimately, such integrated systems could play a key role in advancing carbon circularity, reducing greenhouse gas emissions, and promoting the sustainable management of solid waste resources.

References

- [1] K. Bayramoğlu, M. Bayraktar, A. Seyhan, and O. Yuksel, "Evaluation of techniques to reduce carbon emissions from ships within the scope of revised greenhouse gas emission targets for 2030, 2040, and 2050," *Ocean Engineering*, vol. 334, p. 121605, Aug. 2025, doi: 10.1016/j.oceaneng.2025.121605.
- [2] S. Kim, D. Kong, X. Zheng, and J. H. Park, "Upcycling plastic wastes into value-added products via electrocatalysis and photoelectrocatalysis," *Journal of Energy Chemistry*, vol. 91, pp. 522–541, Apr. 2024, doi: 10.1016/j.jechem.2024.01.010.

- [3] D. Kang, J. Byun, and J. Han, "Electrochemical production of formic acid from carbon dioxide: A life cycle assessment study," *J. Environ. Chem. Eng.*, vol. 9, no. 5, 2021, doi: 10.1016/j.jece.2021.106130.
- [4] A. Francis *et al.*, "Carbon dioxide hydrogenation to methanol: Process simulation and optimization studies," *Int. J. Hydrogen Energy*, vol. 47, no. 86, 2022, doi: 10.1016/j.ijhydene.2022.08.215.
- [5] M. Ramdin *et al.*, "High-Pressure Electrochemical Reduction of CO₂ to Formic Acid/Formate: Effect of pH on the Downstream Separation Process and Economics," *Ind. Eng. Chem. Res.*, vol. 58, no. 51, 2019, doi: 10.1021/acs.iecr.9b03970.
- [6] H. Guzmán *et al.*, "How to make sustainable CO₂ conversion to Methanol: Thermocatalytic versus electrocatalytic technology," *Chemical Engineering Journal*, vol. 417, 2021, doi: 10.1016/j.cej.2020.127973.
- [7] H. Li, H. Cheng, Y. Chen, L. Wang, and J. Yang, "Advances in Solar-Driven, Electro/Photoelectrochemical, and Photothermal-Assisted Upcycling of Waste Plastics," *ACS Appl. Energy Mater.*, vol. 8, no. 7, pp. 4803–4814, Apr. 2025, doi: 10.1021/acsaem.5c00516.
- [8] M. Bazmi, J. Gong, K. Jessen, and T. Tsotsis, "Waste CO₂ capture and utilization for methanol production via a novel membrane contactor reactor process: Techno-economic analysis (TEA), and comparison with other existing and emerging technologies," *Chemical Engineering and Processing - Process Intensification*, vol. 201, p. 109825, Jul. 2024, doi: 10.1016/j.cep.2024.109825.